

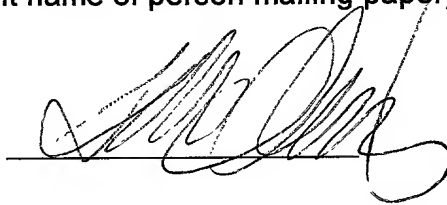
POLYVINYLCHLORIDE PRODUCTS

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I hereby certify that this New Application Transmittal and the documents referred to as enclosed therein are being deposited with the United States Postal Service on this date January 14, 2002 in an envelope as "Express Mail Post Office to Addressee" Mailing Label Number EK206620419US with sufficient postage affixed addressed to the: Assistant Commissioner of Patents and Trademarks, Box New Application Washington, D.C. 20231.

Forrest L. Collins

(Type or print name of person mailing paper)

A handwritten signature in black ink, appearing to read 'Forrest L. Collins', is written over a horizontal line.

BACKGROUND OF THE INVENTION

Cross-Reference to related application. This application claims is a continuation-in-part of Serial number 07/581747 filed September 13, 1990.

1. Field of the invention. This invention relates to the manufacture of polyvinylchloride articles.

2. Description of the art. The manufacture of polyvinylchloride articles comprises a major segment of the plastic materials in use today. In particular, the polyvinylchloride articles with which the present invention is concerned, include siding for homes. To be useful as a siding, polyvinylchloride must exhibit low gloss and have high impact resistance. The impact resistance is important so that the siding will withstand weather. The low gloss is important so that the siding may be painted.

In U.S. Patent 4,336,176 issued June 22, 1982, to Lindner, there is disclosed manufacture of polyvinylchloride compounds utilizing an organo-tin compound and a partial ester having functionality as both an internal lubricant and a co-stabilizer for the plastic. U.S. Patent 4,425,458 issued January 10, 1984 to Lindner describes the use of polyguerbet alcohol esters as lubricants for polycarbonate. U.S. Patent 4,487,874 issued December 11, 1984 to Lindner discloses polycarbonate lubricants which were formed from the partial esterification of pentaerythritol.

U.S. Patent 4,690,977 to Hosoi et al, issued on September 1, 1987, describes vinylchloride polymers comprising a grafted polymer, a vinylchloride polymer, a rubber copolymer, and a lubrication system which includes a wax lubricant and epoxidized soybean oil. Hosoi et al makes similar disclosures in U.S. Patent 4,645,795 issued

February 24, 1987.

Schofield et al in U.S. Patent 4,797,440 issued January 10, 1989 describes a thermoplastic polymeric organic medium in the presence of a lubricant and a processing additive containing a divalent aliphatic radical. The stabilization of polyvinylchloride is taught in U.S. Patent 4,338,226 issued July 6, 1982 to Worschech et al. The stabilization of the vinylchloride polymers is also taught by Crochemore et al in U.S. Patent 4,102,839 issued January 25, 1978. U.S. Patent 3,951,883 issued April 20, 1976 to Ruchlak et al discloses polyvinylchloride molding compositions.

Tenaka et al in U.S. Patent 4,072,657 issued February 7, 1978 discloses synthetic resins which include vinylchloride-vinylacetate copolymers. U.S. Patent 4,392,581 issued July 12, 1983 to Itsubo et al describes vinylchloride resin compositions. U.S. Patent 3,578,621 issued May 11, 1971 to Stapfer describes stearamide waxes which may be utilized for plastic processing.

This invention deals with lubricant additive systems for polyvinylchloride resins which have superior properties with regard to impact resistance and to having low gloss.

Throughout the specification and claims, percentages and ratios are by weight, temperatures are in degrees Celsius, and pressures are in KPa gauge unless otherwise indicated. To the extent that any of the references cited herein are applicable, they are hereby specifically incorporated by reference. Ranges and ratios given herein may be combined.

SUMMARY OF THE INVENTION

A rigid polyvinyl chloride article comprising:

A. a polyvinyl chloride resin;

B. a member selected from the group consisting of:

1. about 0.2 to about 5 weight parts per 100 parts of component A of an ester of a monofunctional organic acid and a dihydric alcohol,
2. about 0.2 to about 5 weight parts per 100 parts of component A of an ester of a monofunctional organic acid and a monohydric alcohol,
3. about 0.2 to about 5 weight parts per 100 parts of component A of a monohydric alcohol ester of a dicarboxylic aromatic acid, and,
4. about 0.2 to about 5 weight parts per 100 parts of component A mixtures of 1 and 2, 1 and 3, 2 and 3, and 1, 2, and 3, and;

C. an organic acid having more than 40 carbon atoms, said organic acid being present in a weight ratio of about 0.2 to about 2 parts per part of component B,

provided further that said article has at least one of the following properties gloss % reflectance @ 60° of less than 65 and withstands a falling dart impact of greater than 247.1 ft-lb/inch..

DETAILED DESCRIPTION OF THE INVENTION

COMPONENT A

Polyvinylchloride resins are made thru the polymerization of vinylchloride monomers. The polyvinylchloride resins obtained are solid materials usually in a particulate form. Polyvinylchloride has the distinct advantage of being able to be extruded into shaped articles which are rigid.

Typically, the polyvinylchloride article is obtained by extruding the polyvinylchloride resin through a twin screw extruder. The extruder masticates the pellets and through the use of heat the polyvinylchloride is obtained as a plastic material. The extruded polyvinylchloride is then fed to a molding or shaping device to obtain the polyvinylchloride article in the desired shape.

During the extrusion of polyvinylchloride, both heat and mechanical energy are applied to the extruder. The mechanical energy input to move the polyvinylchloride through the extruder is eventually present as heat energy in the resin. It is desirable that the polyvinylchloride in the extruder have present materials which function both as internal and external lubricants. An internal lubricant is a material which allows the polyvinylchloride to smoothly flow through the extruder. The smooth flow through the extruder is needed to minimize the energy input to the extruder and to achieve a constant processing rate. The external lubricating component is desirable to avoid having the polyvinylchloride adhere to the inside of the extruder, the die surfaces, or the shaping device.

The manner of masticating the polyvinylchloride is as previously suggested through the use of a twin screw extruder. However, any means may be utilized to

prepare the polyvinylchloride in a plastic state. In particular, the invention utilizes polyvinylchloride pellets having a maximum dimension of 5 mm to 25 mm, preferably 6 mm to 12 mm.

COMPONENT B

The next aspect of the present invention to be discussed is the lubricant. The lubricant B 1 is typically an ester of a monofunctional organic acid and a dihydric alcohol. The monofunctional acid typically contains from 12 to 22, preferably 14 to 20 carbon atoms in the organic acid. The preferred organic acids are saturated materials and in particular stearic acid may be employed herein. The dihydric alcohol typically contains from 2 to 12 carbon atoms and is a saturated material. Preferably the dihydric alcohol contains from 2 to 4 carbon atoms. A preferred example of a dihydric alcohol is ethylene glycol.

Component B 1 is typically fully reacted such that there is little if any unreacted hydroxyl or free acid in the ester. That is, each material used to form an ester in Component B is substantially neutral, preferably at least 90, and more preferably at least 95% of the acid and hydroxyl groups are reacted.

A preferred source of component B 1 is Lubol lubricant available from L&L Industries Chemical, Inc., P.O. Box 281, Montgomeryville, Pennsylvania. A second source of the aforementioned ester is the Henkel Corporation, 300 Brookside Avenue, Ambler, Pennsylvania.

Component B 2 is the ester of a monofunctional organic acid and a monohydric alcohol. Preferably, the aforementioned ester contains from 12 to 22 carbon atoms in the organic acid. More preferably, the organic acid contains from 14 to 20 carbon

atoms and is more preferably a saturated fatty acid. The preferred organic acid for ester B 2 is stearic acid.

The alcohol portion of the ester of component B 2 is typically a saturated fatty alcohol containing from 12 to 22, preferably from 14 to 20 carbon atoms. Most preferably, the ester B 2 is one of stearyl stearate. Stearyl stearate is available from L&L Industries Chemical, Inc. as Lubol 132 lubricant. Stearyl stearate ester is also available as Loxiol G 32 from the Henkel Corporation.

Component B 3 is the monohydric alcohol ester of a dicarboxylic aromatic acid. Typically, the dicarboxylic aromatic acid will be a phthalic acid. The phthalic acid component may be in the form of the ortho, meta, or para form. The monohydric alcohol B 3 will have the same preferred description for the monohydric alcohol B 2. As previously noted, the lubricant B may be employed as mixtures in all proportions of B 1, 2, and 3.

COMPONENT C

The organic acid component of the present invention is utilized as an internal lubricant and metal release agent for polyvinylchloride. The organic acid utilized in the present invention is preferably of a high density and is obtained from oxidized polyethylene. Typical properties useful for the organic acid in the present invention are a bulk density of 34 pounds per cubic foot (0.55 kg/liter). It is also desirable that the organic acid utilized herein has a drop point of about 140^o C and a hardness of less than 0.5. The density is desirably 0.98 g/cc as a solid and the viscosity at 150^o C is 8500 cps. The saponification value is typically 24 and the acid number is typically about 16. The organic typically has 45 to 70, preferably 45 to 60 carbon atoms and is

monofunctional with regard to the carboxylic group.

COMPONENT USAGE

The amount of component B utilized herein is typically about 0.2 to about 5, preferably 0.5 to about 4 weight parts per 100 parts of component A. The amount of the organic acid (component C) is typically at about 0.2 to about 5, preferably about 0.5 to about 3.5 parts per 100 parts of component A. It is also desirable herein that component C is present to component B in a weight ratio of 0.2 to about 2 parts, preferably about 0.3 to about 1.5 parts per part of component B.

ADDITIONAL COMPONENTS

Typically, many additional components are utilized in polyvinylchloride processing. Generally, any of the normally used components in polyvinylchloride processing which do not materially interfere with the functioning of the lubricant system and the desired properties of low gloss and high impact resistance may be utilized herein. For instance, an impact modifier such as Acryloid KM 334 available from Rohm & Haas may be utilized herein at from 1 to 10 weight parts per 100 parts of the polyvinylchloride resin. Fillers and pigments may also be utilized herein typically at from 3 to 20 weight parts per 100 parts of the polyvinylchloride resin. Stabilizers may be utilized herein to neutralize the hydrochloric acid which is generated on degradation of the polyvinylchloride resin. Typically, the stabilizers may be various cadmium, lead, or tin compounds. A preferred tin stabilizer in the present invention is Advisable TM 181. The stabilizers are used in an amount sufficient to neutralize the acid formed upon degradation of the polyvinylchloride. Typically, the stabilizer will be used at 0.3 to 1.0 parts per 100 parts of the polyvinylchloride resin.

A further preferred component herein is a hydrogenated triglyceride. Such materials are utilized as co-lubricants in the processing of the polyvinylchloride in the extruder.

PROCESSING

The components of the present invention are mixed together thoroughly and processed through an extruder. Typically, the extruder will achieve a temperature of 140 to 220, preferably 150 to 200⁰ C during the extrusion process.

The following is a suggested exemplification of the present invention:

TABLE I

	1	2	3	4	5
Polyvinylchloride	100	100	100	100	100
Impact modifier*	5.0	5.0	5.0	5.0	5.0
CaCO ₃ (Superflex 200)	5.0	5.0	5.0	5.0	5.0
TiO ₂ (Titanox 2101)	10.0	10.0	10.0	10.0	10
Process aid Acryloid K120n	0	0	0	0	1.5
Stabilizer (Advastab TM 181)	1.2	1.2	1.2	1.2	1.2
Calcium stearate				1.0	1.0
Organic acid 45 carbon atoms	0.45	0.45	0.45	0.45	
Paraffin (XL 165)					1.0
Ethylene glycol distearate	0.725				
Hydrogenated tallow	0.725				
Stearyl stearate		1.55			
Distearyl phthalate			1.55		
Brabender fusion, mean values					
Fusion time, minutes	2.2	2.2	1.8	2.4	2.3
Maximum torque, m-g	2850	2800	3000	2950	3050
Equilibrium torque, m-g	2650	2750	2700	2950	2850
Stability time, minutes	32	32	27	36	25
Extrusion					
Amps	18.5	19	22	18	16
Die pressure, psi	4100	4550	4800	4100	4100
Output, g/min	103.1	108.3	104.0	112.0	106.5
Product properties					
Impact, ft-lb/inch **	302.7	295.0	265.6	247.1	215.8
Gloss***	45	42	49	65	91.0
Delta E at 15,000 hours	5.2	6.1	5.5	6.7	6.6

The components are shown in parts per hundred parts of the polyvinylchloride. It can be seen that the impact resistance is considerably greater when utilizing the components of the present invention. Similarly, the gloss is significantly lower when utilizing the components of the present invention. Weathering measured by Delta E is also superior.

*Acryloid KM334 ** Impact measured by falling dart test ***Gloss is % reflectance @ 60°